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Prenylated C₆-C₃ compounds from the fruits of *Illicium simonsii* Xian-Fu Wu^a; Yong Li^a; Hai-Ning Lu^a; Shi-Shan Yu^a; Shuang-Gang Ma^a; Jing Liu^a ^a Key Laboratory of Bioactive Substances and Resources Utilization of Chinese Herbal Medicine, Ministry of Education & Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medical College, Beijing, China

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Prenylated C₆-C₃ compounds from the fruits of *Illicium simonsii*

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Two new prenylated C_6-C_3 compounds, 4-*epi*-illicinone E-12-shikimate (1) and 3-hydroxyillifunone B (2), together with five known prenylated C_6-C_3 compounds (3–7), were isolated from the fruits of *Illicium simonsii*. Their structures were elucidated on the basis of extensive spectroscopic methods, including 1D and 2D NMR, CD spectra, and ESI-MS analysis.

Keywords: *Illicium simonsii*; prenylated C_6-C_3 compounds; 4-*epi*-illicinone E-12-shikimate; 3-hydroxyillifunone B

1. Introduction

Prenylated $C_6 - C_3$ compounds, also named phytoquinoids, were isolated frequently from Illicium plants in the previous reports [1–9]. Illicium simonsii belongs to the genus Illicium and is mainly distributed in Guizhou, Sichuan, and Yunnan Provinces of China, India, and Myanmar [10]. Since there are only a few reports concerning the chemical constituents of this plant [11,12], the fruits of I. simonsii were studied as part of an ongoing phytochemical investigation of the genus Illicium. As a result, seven prenylated C_6-C_3 compounds, including two new compounds (1 and 2), were isolated (Figure 1). In this paper, we report the isolation and structural elucidation of the two new compounds.

2. Results and discussion

Compound 1 was obtained as a colorless oil with the molecular formula $C_{22}H_{28}O_9$ established by positive HR-ESI-MS at m/z 437.1817. The IR spectrum showed

absorption bands attributable to hydroxyl groups $(3404 \,\mathrm{cm}^{-1})$, one ester carbonyl $(1708 \,\mathrm{cm}^{-1})$, and one α,β -conjugated carbonyl (1685 cm⁻¹). The ¹H and ¹³C NMR spectra of 1 (Table 1) revealed the presence of an allyl group [$\delta_{\rm H}$ 2.96 (2H, d, $J = 6.5 \text{ Hz}, \text{H}_2\text{-}7), 5.82 (1\text{H}, \text{m}, \text{H}\text{-}8), 5.07$ (1H, dd, J = 10.5, 1.5 Hz, H-9a), and 5.04 (1H, dd, J = 17.5, 1.5 Hz, H-9b); $\delta_{\rm C}$ 33.7 (C-7), 135.8 (C-8), and 117.4 (C-9)], an α,β -conjugated carbonyl group [$\delta_{\rm H}$ 6.63 (1H, s, H-3); $\delta_{\rm C}$ 194.5 (C-1), 138.9 (C-2), and 139.0 (C-3)], two isolated methylene groups [$\delta_{\rm H}$ 3.09 (1H, d, J = 16.5 Hz, H-6a), 2.92 (1H, d, J = 16.5 Hz, H-6b), 2.51 (1H, dd, J = 14.0, 5.0 Hz, H-10a), and 2.23 (1H, dd, J = 14.0, 11.0 Hz, H-10b); $\delta_{\rm C}$ 46.9 (C-6) and 38.4 (C-10)], two dimethyl groups linked to a quaternary carbon [$\delta_{\rm H}$ 1.47 (3H, s, H₃-13) and 1.51 $(3H, s, H_3-14); \delta_C 22.0 (C-13) \text{ and } 22.5 (C-14)$ 14)], and a methylenedioxy group [$\delta_{\rm H}$ 4.92 (1H, s, H-15a) and 5.16 (1H, s, H-15b); $\delta_{\rm C}$ 95.3 (C-15)]. The above NMR spectral

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Figure 1. Structures of compounds 1-7.

data were similar to those of illicinone E [13], indicating that compound 1 possessed a structure moiety (part A), which was supported by HMBC correlations (Figure 2). In addition, the ${}^{1}H$ and ${}^{13}C$ NMR spectral data exhibited the presence of an ester carbonyl group [$\delta_{\rm C}$ 166.2 (C-7')], conjugated double bonds [$\delta_{\rm H}$ 6.61 (1H, br s, H-2'); $\delta_{\rm C}$ 131.0 (C-1') and 138.4 (C-2')], three oxygen-bearing methine groups [$\delta_{\rm H}$ 4.34 (1H, t, J = 4.0 Hz, H-3'), 3.66 (1H, dd, J = 7.0, 4.0 Hz, H-4'), and $3.97 (1H, m, H-5'); \delta_C 66.7 (C-3'), 72.3 (C-$ 4'), and 68.0 (C-5')], and a methene group $[\delta_{\rm H} 2.07 \text{ (1H, dd, } J = 19.0, 5.0 \text{ Hz, H-6'a)}]$ and 2.55 (1H, dd, J = 19.0, 5.0 Hz, H-6'b); $\delta_{\rm C}$ 31.4 (C-6')], which were determined to be a shikimic acid moiety (part B) [14]. In the ¹³C NMR spectrum, the downfield shift of C-11 and C-12 ($\Delta\delta$ + 1.6 and +11.1 ppm, respectively) of part A compared with those of illicinone E suggested that part B was attached to C-12 of part A.

The relative configuration of **1** was determined by the NOE experiment and coupling constant values. Selective irradiation of the signal H-11 caused distinct NOE enhancement of the signal H-15b ($\delta_{\rm H}$ 5.16), suggesting that H-11 and

methylenedioxy were on the same side. No NOEs were observed between H-3' and H-5', suggesting that they were on the different faces of the cyclohexene ring (Figure 2). In the ¹H NMR spectrum, the coupling constant ($J = 7.0 \,\text{Hz}$) between H-4' and H-5' indicated that they were in trans-diaxial disposition in the Haworth projection (Figure 2), which was further supported by the large coupling constant (J = 19.0 Hz) between H-6'a and H-6'b [15]. The negative Cotton effect at 320 nm was in agreement with 4R,5S-configuration [5]. Alkaline hydrolysis of 1 afforded (-)-shikimic acid [14]. Therefore, the structure of 1 was determined as shown in Figure 1.

Compound **2** was obtained as a colorless oil and its HR-ESI-MS exhibited an $[M+Na]^+$ ion at m/z 291.1209 corresponding to the pseudomolecular formula of $C_{14}H_{20}O_5Na$. The IR spectrum showed the presence of hydroxy groups (3429 cm⁻¹), an α , β -conjugated carbonyl (1657 cm⁻¹), and a double bond (1624 cm⁻¹). The ¹H and ¹³C NMR spectral data of **2** indicated the presence of an allyl group [δ_H 2.32 (1H, ddd, J = 5.0, 7.0, 14.0 Hz, H-7a), 2.44 (1H, dd,

No.	1		2	
	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$	$\delta_{\rm H} (J \text{ in Hz})$	$\delta_{\rm C}$
1		194.5		199.2
2		138.9	3.56 m	44.2
3	6.63 s	139.0	4.15 d (3.0)	72.4
4		86.0		78.5
5		111.9		180.4
6	3.09 d (16.5) 2.92 d (16.5)	46.9	5.21 s	96.4
7	2.96 d (6.5)	33.7	2.44 dd (7.5, 14.0) 2 32 ddd (5 0, 7 0, 14 0)	41.8
8	5.82 m	135.8	5.87 m	133.9
9	5.07 dd (10.5, 1.5)	117.4	5.10 dd (10.0, 1.5)	118.2
	5.04 dd (17.5, 1.5)		5.04 dd (17.0, 1.5)	
10	2.51 dd (14.0, 5.0)	38.4	2.36 dd (5.0, 11.5)	26.3
	2.23 dd (14.0, 11.0)		2.10 dd (11.0, 11.5)	
11	4.52 dd (11.0, 5.0)	85.7	4.52 dd (5.0, 11.0)	93.3
12		82.1		70.4
13	1.47 s	22.0	1.27 s	26.5
14	1.51 s	22.5	1.17 s	25.6
15	4.92 s 5.16 s	95.3		
1'		131.0		
2'	6.61 br s	138.4		
3'	4.34 br s	66.7		
4′	3.66 dd (4.0, 7.0)	72.3		
5'	3.97 m	68.0		
6′	2.07 dd (5.0, 19.0) 2.55 dd (5.0, 19.0)	31.4		
7′		166.2		

Table 1. 1 H (500 MHz) and 13 C (125 MHz) NMR spectral data for compounds 1 and 2 in acetone- d_6 .

J = 7.5, 14.0 Hz, H-7b), 5.87 (1H, m, H-8), 5.04 (1H, dd, *J* = 10.0, 1.5 Hz, H-9a), and 5.10 (1H, dd, *J* = 17.0, 1.5 Hz, H-9b); $\delta_{\rm C}$ 41.8 (C-7), 133.9 (C-8), and 118.2 (C-9)], a dimethyl carbinol group [$\delta_{\rm H}$ 1.27 (3H, s, H₃-13) and 1.17 (3H, s, H₃-14); $\delta_{\rm C}$ 70.4 (C-12), 26.5 (C-13), and 25.6 (C-14)], and an α,β-conjugated carbonyl $[\delta_{\rm H}$ 5.21 (1H, s, H-6); $\delta_{\rm C}$ 199.2 (C-1), 180.4 (C-5), and 96.4 (C-6)]. The NMR spectral data of **2** were similar to those of illifunone B [1], except that the signal at $\delta_{\rm C}$ 34.9 (C-3) in illifunone B was shifted downfield to $\delta_{\rm C}$ 72.4 (C-3) in **2**, which disclosed that one proton at C-3 in illifunone B was replaced by a hydroxyl



Figure 2. Structures of parts A and B, key HMBC correlations of 1, and Haworth projection of part B.

group in **2** in combination with their molecular formula.

The relative configuration of 2 was established by the chemical shift of H-11 and the NOE experiment (Figure 3). A synrelationship between the hydroxyl group at C-4 and the dimethyl carbinol group at C-11 was elucidated from the chemical shift of H-11 ($\delta_{\rm H}$ 4.52) appearing at higher field than $\delta_{\rm H}$ 4.6 [7]. Moreover, an *anti*relationship between the allyl group at C-2 and the hydroxyl group at C-3 was confirmed by the observation of NOE for H-3 upon irradiation of H-7b ($\delta_{\rm H}$ 2.44). In addition, a syn-relationship between the allyl group at C-2 and the dimethyl carbinol group at C-11 was established for the NOEs between H-2 and H-11. The absolute configuration of C-2 was determined to be R on the basis of the positive Cotton effect at 310 nm [7]. Therefore, the absolute configuration for compound 2 was deduced as 2R, 3R, 4S, 11R. Thus, the structure of 2 was determined as shown in Figure 1.

The structures of the other five known prenylated C_6-C_3 compounds, 6-allyl-6-(3-methyl-2-butenyl)-3,4-methylenedioxycyclohexa-2,4-dienone (**3**) [3], illicinone E (**4**) [16], illifunone C (**5**) [1,7], illifunone D (**6**) [1,7], and 2,3-dehydroillifunone C (**7**) [7], were elucidated by comparison of their spectral data with those reported.

3. Experimental

3.1 General experimental procedures

Optical rotations were determined on a Perkin-Elmer 241 automatic digital polarimeter. A CD spectrum was obtained from a JOUAN Mark II spectropolarimeter. IR spectra were recorded on a Nicolet 5700 FT-IR spectrometer. 1D and 2D NMR spectra were recorded on a Varian INOVA-500 spectrometer with TMS as an internal standard. ESI-MS were measured on an Agilent 1100 series LC/MSD trap mass spectrometer. HR-ESI-MS spectra were recorded on an Autospec-Ultima ETOF Spec mass spectrometer. Preparative HPLC was peron a Shimadzu LC-6AD formed instrument with an SPD-10A detector. Silica gel GF₂₅₄ for TLC was obtained from Qingdao Marine Chemical Company, Qingdao, China. ODS (50 µ) and Sephadex LH-20 were purchased from Fuji Silysica Chemical Ltd (Greenville, NC, USA).

3.2 Plant material

The fruits of *I. simonsii* were collected from Yunnan Province of China, which was identified by Prof. Lin Ma of the Institute of Materia Medica, Chinese Academy of Medical Sciences and Peking Union Medica, where a voucher specimen (No. 90204) has been deposited.



Figure 3. Key NOE and HMBC correlations of 2.

3.3 Extraction and isolation

The dried powder of the fruits of I. simonsii (1.1 kg) was extracted with 95% EtOH (15 liters \times 3) and concentrated in vacuo to give the crude extract (158 g), which was absorbed by kieselguhr, and then successively extracted with petroleum ether, CHCl₃, EtOAc, and MeOH. The CHCl₃ extract (25 g) was subjected to the ODS column eluted with MeOH-H₂O (from 65:35 to 85:15) to give five fractions (A-E). Fraction B (2.6 g) was subjected to Sephadex LH-20 (CH₂Cl₂-MeOH, 1:1) and then further purified on preparative HPLC (MeOH $-H_2O$, 84:16) to yield compounds 1 (13.1 mg), 2 (9.1 mg), and 7 (16.1 mg). Fraction E (3.3 g) was chromatographed on the ODS column, eluted with a gradient system of MeOH- H_2O (from 60:40 to 90:10), and further separated by preparative HPLC (MeOH- H_2O , 88:12) to afford compounds 3 (4.1 mg), 4 (9.6 mg), 5 (12.4 mg), and 6 (6.1 mg).

3.3.1 Compound 1

Colorless oil; $[\alpha]_{D}^{20} - 38.1$ (c = 0.08, MeOH); UV (MeOH) λ_{max} : 225 nm; CD (MeOH) $\Delta \varepsilon_{200 \text{ nm}}$ 18.2, $\Delta \varepsilon_{224 \text{ nm}}$ 0, $\Delta \varepsilon_{320 \text{ nm}} - 1.66$, $\Delta \varepsilon_{325 \text{ nm}}$ 0; IR (KBr) v_{max} : 3404, 2981, 2907, 1708, 1685, 1613, 1369, 1235, 1137, 1095, 1072 cm⁻¹; ¹H and ¹³C NMR spectral data, see Table 1; positive HR-ESI-MS *m/z*: 437.1817 [M+H]⁺ (calcd for C₂₂H₂₉O₉, 437.1806).

3.3.2 Compound 2

Colorless oil; $[\alpha]_D^{20} - 68.3$ (c = 0.05, MeOH); UV (MeOH) λ_{max} : 265 nm; CD (MeOH) $\Delta \varepsilon_{201 \text{ nm}}$ 0, $\Delta \varepsilon_{256 \text{ nm}} - 11.8$, $\Delta \varepsilon_{296 \text{ nm}}$ 0, $\Delta \varepsilon_{310 \text{ nm}}$ 1.04, $\Delta \varepsilon_{348 \text{ nm}}$ 0; IR (KBr) ν_{max} : 3429, 2988, 2911, 1657, 1624, 1389, 1376, 1190, 1079 cm⁻¹; ¹H and ¹³C NMR spectral data, see Table 1; positive HR-ESI-MS *m*/*z*: 291.1209 [M+Na]⁺ (calcd for C₁₄H₂₀O₅Na, 291.1203).

3.4 Alkaline hydrolysis of 1

Compound 1 (6.2 mg) was hydrolyzed with 0.5 M NaOH (3 ml) and MeOH (0.5 ml) for 1 h at room temperature. After adjusting the pH to 5.0 with 0.5 M HCl, the reaction mixture was extracted with EtOAc, which was separated by HPLC (CH₃CN-H₂O, 18:92) to afford (–)-shikimic acid (1.2 mg).

3.4.1 (-)-Shikimic acid

Colorless powder; $[\alpha]_D^{20} - 138.1$ (*c* = 0.05, MeOH); ¹H NMR (500 MHz, acetone-*d*₆): $\delta_{\rm H}$ 6.71 (1H, m, H-2), 4.40 (1H, t, *J* = 4.0 Hz, H-3), 3.68 (1H, dd, *J* = 7.0, 4.0 Hz, H-4), 3.99 (1H, m, H-5), 2.68 (1H, dd, *J* = 19.0, 5.0 Hz, H-6a), 2.11 (1H, dd, *J* = 19.0, 5.0 Hz, H-6b); ¹³C NMR (125 MHz, acetone-*d*₆): $\delta_{\rm C}$ 138.1 (C-1), 133.7 (C-2), 71.6 (C-3), 68.1 (C-4), 67.2 (C-5), 31.3 (C-6), 170.1 (C-7).

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